Acta Crystallographica Section C Crystal Structure Communications

ISSN 0108-2701

trans-Carbonyliodotris(triphenylstibine-*kSb*)rhodium(I)

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Received 29 August 2002 Accepted 11 October 2002 Online 8 November 2002

The crystal structure of the title compound, $[RhI(C_{18}H_{15}Sb)_{3}-(CO)]$, represents a rare example of a crystallographically characterized five-coordinate Rh^{I} -SbPh₃ complex. The compound crystallizes with the I-Rh-CO core on a threefold rotation axis, with three crystallographically equivalent triphenylstibine ligands. Selected geometric parameters are: Rh-I = 2.7159 (8), Rh-Sb = 2.5962 (4), $Rh-C_{CO} = 1.825$ (6) and C_{CO} -O 1.153 (6) Å, and Sb-Rh-I = 89.374 (10) and $Sb-Rh-C_{CO} = 90.626$ (10)°. The cone angle of the SbPh₃ ligand was determined as 137°, according to the Tolman model.

Comment

Although stibine complexes of Rh have been known since the 1950s (Vallarino, 1957), surprisingly few crystallographic studies of these complexes have been reported to date. As has been pointed out in earlier investigations (Ugo *et al.*, 1969; Otto & Roodt, 2002), several experimental problems may arise during the characterization of these complexes, making X-ray crystallography the method of choice. As part of our systematic investigation of these systems (Otto *et al.*, 2002; Otto & Roodt, 2002), we obtained single crystals of *trans*-carbonyliodotris(triphenylstibine- κSb)rhodium(I), (I), and the crystal structure is presented here.



Compound (I) (Fig. 1) crystallizes with an almost perfect trigonal-bipyramidal geometry in the trigonal space group $P\overline{3}$. The packing in the crystal is governed by van der Waals forces alone; no significant intermolecular interactions are observed. The OC-Rh-I core is situated along the threefold rotation axis, resulting in three crystallographically equivalent SbPh₃

ligands around the Rh atom with angles of exactly 120° . Accordingly, the OC-Rh-I core is exactly linear and the C1-Rh-Sb and I-Rh-Sb angles are close to 90° , at 90.626 (10) and 89.374 (10)°, respectively.

An I-Rh-Sb-C11 torsion angle of 175.77 (10)° indicates phenyl ring 1 (C11-C16) points almost directly towards the CO moiety, whereas rings 2 (C21-C26) and 3 (C31-C36) are staggered with respect to the Rh-I bond, with I-Rh-Sb-C21 and I-Rh-Sb-C31 torsion angles of -63.88 (11) and 57.89 (9)°, respectively. These specific geometric orientations may account for the observation that the SbPh₃ ligands are marginally displaced towards the larger I ligand.

Based on molecular models, Tolman (1977) predicted that increasing the M-L or L-C (M is a metal and L is a ligand) bond lengths by 0.1 Å (for ligands containing donor atoms other than P) would result in a 3–5° decrease in the cone angle of the ligand. In this regard, the effective cone angle, as defined earlier (Otto *et al.*, 2000), utilizes the observed Rh–Sb bond distance, and the cone angle for SbPh₃ in (I) was calculated as 137°. This value is in excellent agreement with both Tolman's prediction and the average of 139° for six independent SbPh₃ ligands obtained during a previous study (Otto & Roodt, 2002).

In Table 2, compound (I) is compared with related *trans*- $[Rh(X)CO(LPh_3)_n]$ (n = 2 or 3; L is Sb or P) complexes, illustrating the effect of different X ligands on the geometric parameters of four- and five-coordinate SbPh₃ complexes of Rh^I. From these data, it is clear that the Rh—Sb bond length of 2.5962 (4) Å in (I) is comparable with the value of



Figure 1

A view of the structure of (I), showing the atom-numbering scheme and displacement ellipsoids at the 30% probability level. H atoms have been omitted for clarity. The complex is numbered with the first digit of the phenyl rings referring to the number of the ring (1–3) and the second digit referring to the number of the atom within the ring (1–6) [symmetry codes: (i) -x + y, 1 - x, z; (ii) 1 - y, x - y + 1, z].

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2.5981 (5) Å found in the analogous five-coordinate Cl complex, but is slightly longer than the values of 2.568 (2) Å in the acetyl complex and 2.5655 (2) Å in the four-coordinate Cl complex. Not many Rh–I bond distances are known for complexes of this kind, but the Rh–I bond length of 2.7159 (8) Å in (I) is considerably longer than the value of 2.683 (1) Å determined for *trans*-[RhI(CO)(PPh₃)₂] (Basson *et al.*, 1990). This may be a result of the increased steric crowding in the five-coordinate complex or the increased electron density introduced by the three SbPh₃ ligands. A similar observation was made in relation to the Rh–Cl bond distances of the four- and five-coordinate triphenylstibine complexes (Otto & Roodt, 2002) listed in Table 2, where an elongation from 2.315 (3) to 2.4094 (18) Å was induced by the increase in coordination number.

Experimental

NaI (17 mg, 0.113 mmol) was added to a nitrogen-flushed solution of the [Rh(μ -Cl)(CO)₂]₂ dimer (20 mg, 0.051 mmol) in acetone (5 ml). The reaction medium immediately took on a deep-red colour and was stirred for a further 5 min after addition. Stirring was then discontinued and SbPh₃ (126 mg, 0.357 mmol) dissolved in acetone (7 ml) was added carefully so as to avoid the least disturbance to the solution. Deep-red rectangular crystals of (I) soon started separating from the solution; yields > 80% based on Rh were obtained. Spectroscopic analysis, IR (KBr, ν): 1977 cm⁻¹ (CO); IR (dichloromethane, ν): 1978 cm⁻¹ (CO).

Crystal data

[RhI(C ₁₈ H ₁₅ Sb) ₃ (CO)]	Mo $K\alpha$ radiation
$M_r = 1316.97$	Cell parameters from 4642
Trigonal, $P\overline{3}$	reflections
a = 14.462 (2) Å	$\theta = 2.2-24.8^{\circ}$
c = 13.902 (3) Å	$\mu = 2.56 \text{ mm}^{-1}$
V = 2518.2 (7) Å ³	T = 293 (2) K
Z = 2	Parallelepiped, dark red
$D_x = 1.737 \text{ Mg m}^{-3}$	$0.29 \times 0.10 \times 0.08 \text{ mm}$
Data collection	
Siemens SMART CCD area-	5324 independent reflections
detector diffractometer	2651 reflections with $I > 2\sigma(I)$
ω scans	$R_{\rm int} = 0.094$
Absorption correction: multi-scan	$\theta_{\rm max} = 31.8^{\circ}$

Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{min} = 0.493, T_{max} = 0.815$ 24 940 measured reflections

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.031$	$w = 1/[\sigma^2 (F_o^2) + (0.0154P)^2]$
$wR(F^2) = 0.056$	where $P = (F_o^2 + 2F_c^2)/3$
S = 0.86	$(\Delta/\sigma)_{\rm max} = 0.003$
5324 reflections	$\Delta \rho_{\rm max} = 0.43 \ {\rm e} \ {\rm \AA}^{-3}$
185 parameters	$\Delta \rho_{\rm min} = -0.67 \ {\rm e} \ {\rm \AA}^{-3}$

 $h = -18 \rightarrow 21$

 $k = -20 \rightarrow 20$

 $l=-20\rightarrow 20$

H atoms were treated as riding, with C-H = 0.93 Å. Even though the structure contains solvent-accessible voids of 115 Å³, both the minimum and maximum residual electron-density peaks were smaller than 1 e Å⁻³, indicating that no molecular fragments remain unaccounted for.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SAINT* (Siemens, 1995); data reduction: *SAINT*; program(s) used to solve

Table 1

Selected geometric parameters (Å, °).

Rh-C1	1.825 (6)	Sb-C21	2.143 (3)
Rh-Sb	2.5962 (4)	Sb-C31	2.129 (3)
Rh-I	2.7159 (8)	C1-O	1.153 (6)
Sb-C11	2.140 (3)		
C1-Rh-Sb	90.626 (10)	C21-Sb-C31	96.21 (12)
Sb-Rh-I	89.374 (10)	C11-Sb-Rh	112.73 (9)
C11-Sb-C21	99.57 (12)	C21-Sb-Rh	124.39 (8)
C11-Sb-C31	101.04 (12)	C31-Sb-Rh	118.84 (8)
I-Rh-Sb-C11 I-Rh-Sb-C21	175.77 (10) -63.88 (11)	I-Rh-Sb-C31	57.89 (9)

Table 2

Comparative X-ray data for trans- $[Rh(X)CO(LPh_3)_n]$ complexes.

X	n	L	Rh-C1 (Å)	Rh-L (Å)	$\operatorname{Rh}-X(\operatorname{\AA})$	Reference
Cl	2	Sb	1.797 (13)	2.5655 (2)	2.315 (3)	а
Cl	3	Sb	1.875 (7)	2.5981 (5)	2.4094 (18)	а
I	2	Р	1.81 (1)	2.326 (2)	2.683 (1)	b
Ι	3	Sb	1.825 (6)	2.5962 (4)	2.7159 (8)	с
COCH ₃	3	Sb	1.91 (2)	2.568 (2)	2.095 (16)	d

References: (a) Otto & Roodt (2002); (b) Basson et al. (1990); (c) this work; (d) Lamprecht et al. (1986).

structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *SHELXL*97.

Financial assistance from the South African NRF, and from the Research Funds of the Rand Afrikaans University and the University of the Free State, is gratefully acknowledged. The authors thank the Chemical Centre of the University of Lund, Sweden, for the use of their diffractometer for the data collection.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: JZ1526). Services for accessing these data are described at the back of the journal.

References

Basson, S. S., Leipoldt, J. G. & Roodt, A. (1990). Acta Cryst. C46, 142-143.

- Brandenburg, K. (1999). *DIAMOND*. Release 2.1c. Crystal Impact GbR, Bonn, Germany.
- Lamprecht, G. J., Van Biljon, C. P. & Leipoldt, J. G. (1986). Inorg. Chim. Acta, 119, L1–L4.
- Otto, S., Mzamane, S. N. & Roodt, A. (2002). Modern Coordination Chemistry, The Legacy of Joseph Chatt, edited by G. J. Leigh & N. Winterton, pp. 328– 339. London: Royal Society of Chemistry.
- Otto, S. & Roodt, A. (2002). Inorg. Chim. Acta, 331, 199-207.
- Otto, S., Roodt, A. & Smith, J. (2000). Inorg. Chim. Acta, 303, 295-299.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Siemens (1995). SMART and SAINT. Versions 4.0. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Tolman, C. A. (1977). Chem. Rev. 77, 313–348.
- Ugo, R., Bonati, F. & Cenini, S. (1969). Inorg. Chim. Acta, 3, 220–222.
- Vallarino, L. (1957). J. Chem. Soc. pp. 2287-2292.